

Access to Indole Derivatives from Diaryliodonium Salts and 2-Alkynylanilines

Pengfei Li, Yunxiang Weng, Xianxiang Xu, and Xiuling Cui*

Key Laboratory of Xiamen Marine and Gene Drugs, Institutes of Molecular Medicine and School of Biomedical Sciences, Huaqiao University & Engineering Research Center of Molecular Medicine, Ministry of Education, Xiamen 361021, China

Supporting Information



ABSTRACT: An efficient, environmentally friendly, and operationally simple procedure to 1,2-disubstituted indoles from 2alkynylanilines and diaryliodonium salts has been developed. This reaction proceeds smoothly under metal-free conditions. The products obtained could be transferred into 3,3'-diindolylmethane with DMSO catalyzed by palladium. The isotopic label experiments indicated that the methylene group in 3,3'-diindolylmethane is derived from DMSO. The diverse indoles were obtained in up to 90% yield for 28 examples.

INTRODUCTION

Indole is an important motif and widely exist in natural products, advanced functional materials, and bioactive molecules.¹ Consequently, developing an efficient protocol for the synthesis of indole and its derivatives has received tremendous interest during the past decades.² Classical routes include Fischer synthesis,³ Batcho-Leimgruber synthesis, Gassman synthesis, Madelung cyclization of N-acyl-o-toluidines, and the reductive cyclization of o-nitrobenzyl ketones.⁴ Recently, transition-metal-catalyzed transformations and multicomponent reactions have also been widely applied in indole synthesis.⁵⁻ Although these methods have contributed greatly to this area, there are still some respective limitations, such as harsh reaction conditions and the required transition metal, that may cause the potential contamination of the products, which limits their applications, especially in the pharmaceutical industry. Therefore, development of a green, metal-free, and convenient method is highly desirable. Diaryliodonium salt is an important electrophilic arylating reagent in organic synthesis due to its low toxicity, easy handling, and excellent selectivity.⁸ In our continuing interest in hypervalent iodine,⁹ we developed a novel protocol to 1,2-disubstituted indoles from 2-alkynylanilines with diaryliodonium salts.

RESULTS AND DISCUSSION

We initially studied the reaction of 2-alkynylanilines 1a and diphenyliodonium hexafluorophosphate 2a in 1,2-dichloroethane (DCE) at room temperature. The desired product 3a was obtained in 25% yield (Table 1, entry 1). Inspired by this result, various solvents were investigated. We found that DMSO, toluene, MeOH, EtOH, and H₂O were ineffective in this reaction (Table 1, entries 2–5 and 7). Interestingly, *t*-BuOH showed optimal influence on this transformation and Table 1. Optimization of the Reaction Conditions for the Reaction of 2-Alkynylanilines 1a and Diphenyliodonium Hexafluorophosphate $2a^{a}$

	Ph		
	+ Ph-	$\stackrel{i}{\vdash}_{Ph}$ solvent, temp $\stackrel{-}{PF_6}$	Ph Ph
	Ia	Za	Ja
entry	solvent	T (°C)	yield (%) ^b
1	DCE	rt	25
2	DMSO	rt	27
3	toluene	rt	39
4	MeOH	rt	6
5	EtOH	rt	13
6	t-BuOH	rt	48
7	H_2O	rt	<5
8	t-BuOH	50	56
9	t-BuOH	60	69
10	t-BuOH	70	70
11 ^c	t-BuOH	60	78
12 ^d	t-BuOH	60	78
13 ^e	t-BuOH	60	77
14 ^f	t-BuOH	60	85

^{*a*}Reaction conditions: **1a** (0.2 mmol), **2a** (0.2 mmol), solvent (1 mL) under air at room temperature. ^{*b*}Isolated yields. ^{*c*}*t*-BuOH (2 mL). ^{*d*}Under N₂. ^{*c*}Under O₂. ^{*f*}Cu(OTf)₂ (0.1 equiv).

afforded the desired product **3a** in 48% yield (Table 1, entry 6). Then, the reaction temperature was examined (Table 1, entries 8-10). The best result was obtained at 60 °C (Table 1, entry

Received: January 16, 2016 Published: May 8, 2016

Table 2. Scope of 2-Alkynylanilines 1^a



^aReaction conditions: 1 (0.2 mmol) and 2a (0.2 mmol) in t-BuOH (2 mL) under air at 60 °C for 18–45 h.

9). When the solvent volume was increased to 2 mL from 1 mL, the yield of **3a** could be further improved to 78% (Table 1, entry 11). The yield of **3a** was not significantly improved under O_2 or N_2 atmosphere (Table 1, entries 12 and 13). The presence of catalyst Cu did not significantly improve this reaction (Table 1, entry 14). Finally, the optimal reaction conditions were identified as follows: **1a** (0.2 mmol) and **2a** (0.2 mmol) in *t*-BuOH (2 mL) under air at 60 °C.

With the optimized reaction conditions in hand, we explored the substrate scope of 2-alkynylanilines 1 first (Table 2). The electron density of the substituent R² had no significant influence on this reaction. R^2 can be phenyl. Electron-donating (p-Me, p-OMe, and m-Me) and electron-withdrawing (p-F, p-P)Cl, *p*-Br, and *p*-NO₂) groups on the phenyl rings could be welltolerated. The desired products were obtained in moderate to good yields (Table 2, 3a-h). The electronic property of the R¹ also had no obvious influence. When $R^1 = Me$, F, and Cl, the corresponding products were obtained in 78%, 81%, and 71% yields, respectively (Table 2, 3i-k). A heteroaryl group was tolerated as well. The corresponding product was obtained in 81% yield for thiophene (Table 2, 31). Both CO₂R- and alkylsubstituted alkynes were tested under the standard conditions. No reaction occurred for CO2Et-substituted alkyne, and a trace amount of the desired product (10% yield) was obtained for npentylalkyne.

Next, the scope of diaryliodonium salts was explored. The electron density of the substituent of the diaryliodonium salts **2** (*p*-Me, *p*-t-Bu, *p*-F, *p*-Cl, *p*-Br, *p*-I, *p*-CO₂Et) have no obvious influence on the transformation. The products **3ba**–**3ha** were obtained in good yields (Table 3, entries 1–7). The steric hindrance of the substituents had slight effect on the yields. 2-Phenyl-1-(*p*-tolyl)-1*H*-indole **3ba** and 2-phenyl-1-(*o*-tolyl)-1*H*-indole **3ba** and 71% yield, respectively (Table 3, entry 8 vs entry 1). When unsymmetric diaryliodonium salts **2** were used as substrates, *N*-arylation of indoles is sensitive to the electron density and steric hindrance. *N*-(*p*-Ior-F-phenyl) indole was obtained as a main product because of the electron-withdrawing ability of I or F (Table 3, entries 5)

and 6). However, no byproducts were obtained for 2h-2j, owing to the large steric hindrance (Table 3, entries 7–9).

Indole [1,2-f] phenanthridines is widely applied in advanced materials.^{10,11} Under the standard conditions, the reaction of **1m** (0.2 mmol) with **2a** (2 equiv) could provide the corresponding product **3m** in 81% yield (eq 1).



To demonstrate the scalability of the method, a gram-scale experiment was conducted with 10 mmol (1.93 g) of 1a, 10 mmol (4.26 g) of 2a, and *t*-BuOH (100 mL) (Scheme 1). The product 3a was obtained in 57% yield (1.53 g) (eq 2).

To further extensive application of our methodology, the synthesis of bis(1,2-diphenyl-1*H*-indol-3-yl)methane from titled products has been explored. The reaction of **3a** with DMSO afforded bis(1,2-diphenyl-1*H*-indol-3-yl)methane (**4a**) in 47% yield in the presence of Pd(OAc)₂ (0.1 equiv), AgOAc (2 equiv), and DMSO (2 mL) under N₂ at 100 °C. Inspired by this result, the reaction conditions were optimized. The optimized reaction conditions were identified as follows: **3a** (0.05 mmol), Pd(OAc)₂ (0.1 equiv), AgOAc (1 equiv), DMSO (1 mL), 130 °C under N₂ atmosphere (Table 4). The products of **4a**-**f** could be obtained in good to excellent yields (Table 5).

To gain insight into the mechanism of this process, four control experiments were conducted (Scheme 2). Under the standard reaction conditions, the reaction of a1, 1d, and 2a gave 3d in 76% yield in t-BuOH (2 mL) at 60 °C (Scheme 2, eq 3). Compound a1 did not participate in the reaction. Meanwhile, the reaction of 2-phenylindole with diphenyliodonium hexafluorophosphate (2a) did not give the desire product under the standard reaction conditions (Scheme 2, eq 4). These results suggested that both a1 and 2-phenylindole as key intermediates were ruled out in this procedure. When DMSO-

Table 3. Scope of Diaryliodonium Salts 2^{a}

	Ph				
		$Ar^2 \xrightarrow{tert-BuOH}_{60 °C, air}$		—Ph	
	√ ` _{NH₂} 01	T	Ar	.1	
entry	substrate 2	product 3	5	yield (%)	
1	2b	N N	3ba	80	
2	t-Bu 2c Bu-t	Ph	3ca	84	
3	CI Zd		3da	75	
4	Br 2e Br	Cl Ph N	3ea	69	
5		Br Ph	3fa/3a	79/12	
6	2g	Ph N	3ga/3a	65/15	
7	2h		3ha	77	
8			₂Et 3ia	71	
9	i-Pr i-Pr i-Pr i-Pr i-Pr i-Pr i-Pr i-Pr	Ph N	3ja Br	76	

^aReaction conditions: 1a (0.2 mmol) and 2 (0.2 mmol) in *t*-BuOH (2 mL) under air at 60 °C.

Scheme 1. Gram-Scale Experiment



 d_6 and ¹³C-labeled DMSO were used as solvent under the standard reaction conditions, **4aa** and **4ab** were provided. The corresponding structures were confirmed by ¹H and ¹³C NMR spectra (Scheme 2, eqs 5 and 6). These results indicated that the methylene carbon atom in the product **4a** was derived from the methyl group of DMSO.

According to results obtained above, a plausible reaction mechanism was proposed and is shown in Scheme 3. In part A,

Table 4. Optimization of the Reaction Conditions for the Reaction of 1,2-Diphenyl-1H-indole 3 and DMSO^a



^{*a*}Reaction conditions: **3a** (0.05 mmol), Pd(OAc)₂ (0.1 equiv), oxidant (1 equiv), DMSO (1 mL), N₂, 80 °C. ^{*b*}Isolated yields. ^{*c*}100 °C. ^{*d*}130 °C. ^{*c*}Under O₂.

electrophic attack of diaryliodonium salts 2 at the N atom from 2-(phenylethynyl)aniline (1) gave the intermediate I, which then transformed into intermediate II. The intermediate II was detected by NMR (Figure SIA in Supporting Information) and Q-TOF HRMs (Figure SIB in Supporting Information) and finally converted into the 1,2-diphenyl-1*H*-indole 3 as well as PhI. In part B, the C3–H bond of 1,2-diphenyl-1*H*-indole 3 was activated by Pd(OAc)₂, affording a palladated intermediate III. The palladium atom in intermediate III might coordinate with DMSO and gave intermediate IV. Reductive elimination of IV produced the intermediate V (IV and V were detected by

Table 5. Scope of 1,2-Diphenyl-1H-indole 3^a

Q-TOF HRMs, as seen in Figures SIC and SID in the Supporting Information). The final product 4 was finally obtained through 1,4-addition of 1,2-diphenyl-1*H*-indole 3 with the intermediate V.

CONCLUSION

In summary, we have presented an environmentally friendly, efficient, and operationally simple protocol to synthesize 1,2disubstituted indole derivatives using diaryliodonium salts and 2-alkynylanilines as starting materials. Compared with the previous methods, this method is not sensitive to air and is carried out under metal-free conditions. The products could be transformed into bis(1,2-diphenyl-1H-indol-3-yl)methane with DMSO catalyzed by palladium.

EXPERIMENTAL SECTION

General Information. All commercial materials and solvents were used directly without further purification. Melting points were determined in open glass capillaries and were uncorrected. All the compounds were characterized by IR, and only major peaks were reported in cm⁻¹. ¹H NMR spectra were recorded on 400 MHz spectrometers, and ¹³C NMR spectra were recorded on a 100 MHz spectrometer. Chemical shifts (in ppm) were referenced to tetramethylsilane ($\delta = 0$ ppm) in CDCl₃ as an internal standard at room temperature. ¹³C NMR spectra were obtained by using the same NMR spectrometer and were calibrated with CDCl₃ ($\delta = 77.00$ ppm). High-resolution mass spectra (HRMS) were equipped with an ESI source and a TOF detector. Column chromatography was performed on silica gel (70–230 mesh ASTM) using the reported eluents. Thinlayer chromatography (TLC) was carried out on 4 × 15 cm plates with a layer thickness of 0.2 mm (silica gel 60 F254).

Preparation of 2-Alkynylanilines 1. Synthesis of 2-alkynylanilines 1 was accomplished by a modified Sonogashira¹² coupling reaction of the 2-iodoanilines and alkynes as follows: 2-iodoaniline (1 equiv), alkyne (1 equiv), $Pd(PPh_3)_2Cl_2$ (0.02 equiv), CuI (0.02 equiv), and Et_3N (0.2 M) were stirred under N₂ at 60 °C until TLC indicated



^aReaction conditions: 3 (0.05 mmol), Pd(OAc)₂ (0.1 equiv), and AgOAc (1 equiv) in DMSO (1 mL) under N₂ at 130 °C for 20 h.

Article

Scheme 2. Control Experiments



Scheme 3. Plausible Reaction Mechanism



complete consumption (for 24-48 h). Then water was added to the solution and the mixture stirred for 0.5 h at room temperature. The organic layer was concentrated under reduced pressure to give a residue, which was purified by flash chromatography (silica gel, eluent EtOAc/PE = 1/50) to give 1.

Preparation of Diaryliodonium Salts 2. 3-Chloroperoxybenzoic acid (2.6 mmol) and aryl iodide were dissolved in CH_2Cl_2 (10 mL) in a sealed tube. The arene (2.6 mmol) was added. The solution was cooled to 0 °C. Then TfOH was added dropwise. The reaction was

stirred at the indicated temperature for 0.2–22 h. The solvent was concentrated under vacuum. Et₂O (10 mL) was added and the mixture was stirred at room temperature for 10 min to precipitate out an off-white solid. To ensure complete precipitation, the flask was stored in the freezer for 30 min before the solid was filtered off, washed with Et₂O, and dried under vacuum to give diaryliodonium salts 2.¹³

Preparation of 1,2-Diphenyl-1*H***-indoles 3.** A mixture of diaryliodonium salts 2 (0.2 mmol, 1 equiv) and 2-alkynylanilines 1 (0.2 mmol, 1 equiv) in *t*-BuOH (2 mL) was stirred at 60 °C in a tube

under air atmosphere for 18-45 h until TLC indicated complete consumption. Then the solvent was evaporated under reduced pressure and the residue was purified by chromatography (silica gel, eluent EtOAc/PE = 1/40) to give 3.

Intermediate II. ¹H NMR (400 MHz, $CDCl_3$) δ 9.76 (s, 1H), 8.24–6.21 (m, 20H), 2.21 (s, 3H).

1,2-Diphenyl-1H-indole (**3a**).^{14a} Yield 42 mg (78%) of a white solid; mp 79–80 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.78–7.73 (m, 1H), 7.50–7.44 (m, 2H), 7.42–7.28 (m, 9H), 7.27–7.22 (m, 2H), 6.88 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 140.7 (s), 139.0 (s), 138.5 (s), 132.5 (s), 129.2 (s), 128.9 (s), 128.1 (t, *J* = 11.2 Hz), 127.2 (d, *J* = 9.6 Hz), 122.3 (s), 120.6 (d, *J* = 17.6 Hz), 110.6 (s), 103.7 (s). 1-Phenyl-2-(p-tolyl)-1H-indole (**3b**).^{14b} Yield 35 mg (62%) of a

1-Phenyl-2-(p-tolyl)-1H-indole (**3b**).¹⁴⁰ Yield 35 mg (62%) of a yellow solid; mp 129–131 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.75–7.71 (m, 1H), 7.47 (t, *J* = 7.5 Hz, 2H), 7.42–7.37 (m, 1H), 7.34–7.28 (m, 3H), 7.24–7.19 (m, 4H), 7.10 (d, *J* = 8.1 Hz, 2H), 6.83 (s, 1H), 2.36 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 140.9(s), 138.9 (s), 138.6 (s), 137.1 (s), 129.6 (s), 129.2 (s), 128.9 (s), 128.8 (s), 128.3 (s), 128.1 (s), 127.1 (s), 122.1 (s), 120.6 (s), 120.4 (s), 110.5 (s), 103.2 (s), 21.2 (s).

2-(4-Methoxyphenyl)-1-phenyl-1H-indole (3c).^{15a} Yield 35 mg (59%) of a white solid; mp 117–119 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.74–7.70 (m, 1H), 7.46 (t, J = 7.4 Hz, 2H), 7.40 (d, J = 7.3 Hz, 1H), 7.34–7.28 (m, 3H), 7.23 (ddd, J = 8.4, 5.5, 1.4 Hz, 4H), 6.86–6.77 (m, 3H), 3.82 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 158.9 (s), 140.6 (s), 138.8 (s), 138.6 (s), 130.1 (s), 129.2 (s), 128.3 (s), 128.1 (s),127.1 (s), 125.1 (s), 122.0 (s), 120.6 (s), 120.3 (s), 113.6 (s), 110.5 (s), 102.8 (s), 55.2 (s).

2-(4-Fluorophenyl)-1-phenyl-1H-indole (**3d**).^{15b} Yield 42 mg (74%) of a yellow solid; mp 125–127 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.73 (dd, J = 6.0, 3.0 Hz, 1H), 7.50–7.45 (m, 2H), 7.43–7.38 (m, 1H), 7.34 (dd, J = 6.0, 3.3 Hz, 1H), 7.31–7.22 (m, 7H), 7.02–6.96 (m, 2H), 6.82 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 162.1 (d, J_{C-F} = 247.5 Hz), 139.6 (s), 138.8 (s), 138.2 (s), 130.5 (d, J_{C-F} = 8.1 Hz), 129.3 (s), 128.6 (d, J_{C-F} = 3.3 Hz), 128.3 (s), 128.2 (s), 127.3 (s), 122.4 (s), 120.8 (s), 120.5 (s), 115.3 (s), 115.2 (d, J_{C-F} = 21.6 Hz), 110.6 (s), 103.5 (s).

2-(4-Chlorophenyl)-1-phenyl-1H-indole (**3e**).^{15a} Yield 45 mg (75%) of a yellow solid; mp 179–181 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.76–7.69 (m, 1H), 7.47 (t, *J* = 7.6 Hz, 2H), 7.41 (t, *J* = 7.3 Hz, 1H), 7.33 (dd, *J* = 5.5, 3.2 Hz, 1H), 7.30–7.26 (m, 3H), 7.25–7.21 (m, 5H), 6.84 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 139.4 (s), 139.1 (s), 138.3 (s), 133.3 (s), 131.0 (s), 130.0 (s), 129.4 (s), 128.4 (s), 128.1 (s), 128.0 (s), 127.4 (s), 122.6 (s), 120.9 (s), 120.6 (s), 110.6 (s), 104.0 (s).

2-(4-Bromophenyl)-1-phenyl-1H-indole (**3f**).^{18b} Yield 50 mg (72%) of a yellow solid; mp 175–176 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.75–7.71 (m, 1H), 7.50–7.45 (m, 2H), 7.42 (ddd, J = 6.7, 3.5, 1.6 Hz, 3H), 7.33 (dd, J = 6.1, 3.3 Hz, 1H), 7.30–7.27 (m, 2H), 7.25–7.22 (m, 2H), 7.18–7.15 (m, 2H), 6.85 (d, J = 0.5 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 139.4 (s), 139.1 (s), 138.2 (s), 131.4 (s), 131.3 (s), 130.3 (s), 129.4 (s), 128.1 (s), 128.0 (s), 127.4 (s), 122.6 (s), 121.5 (s), 120.9 (s), 120.6 (s), 110.6 (s), 104.0 (s).

2-(4-Nitrophenyl)-1-phenyl-1H-indole (**3g**). Yield 50 mg (80%) of a yellow solid; mp 138–140 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.15–8.10 (m, 2H), 7.75 (dd, J = 6.6, 1.8 Hz, 1H), 7.53–7.41 (m, 5H), 7.35–7.24 (m, 5H), 7.01 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 146.4 (s), 139.8 (s), 138.9 (s), 137.9 (s), 129.7 (s), 128.9 (s), 127.9 (s), 127.8 (s), 123.7 (s), 123.5 (s), 121.3 (s), 121.1 (s), 110.9 (s), 106.2 (s); HRMS (ESI) *m/z* calcd for C₂₀H₁₄N₂O₂ [M + H]⁺ 315.1134, found 315.1133.

1-Phenyl-2-(m-tolyl)-1H-indole (**3h**).^{15c} Yield 46 mg (82%) of a yellow solid; mp 107–109 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.78–7.73 (m, 1H), 7.48 (t, *J* = 7.4 Hz, 2H), 7.41 (dd, *J* = 5.8, 3.8 Hz, 1H), 7.39–7.35 (m, 1H), 7.35–7.31 (m, 2H), 7.25 (dt, *J* = 7.4, 4.0 Hz, 3H), 7.17 (t, *J* = 7.6 Hz, 1H), 7.12–7.06 (m, 2H), 6.87 (s, 1H), 2.34 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 140.9 (s), 139.0 (s), 138.6 (s), 137.7 (s), 132.4 (s), 129.6 (s), 129.2 (s), 128.3 (s), 128.0 (s), 127.9 (s), 127.1 (s), 126.0 (s), 122.2 (s), 120.6 (s), 120.5 (s), 110.6 (s), 103.5 (s), 21.4 (s).

5-Methyl-1,2-diphenyl-1H-indole (3i).^{16a} Yield 44 mg (78%) of a yellow solid; mp 91–93 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.51 (s, 1H), 7.47–7.41 (m, 2H), 7.38 (dd, *J* = 4.9, 3.7 Hz, 1H), 7.30–7.21 (m, 8H), 7.04 (dd, *J* = 8.4, 1.3 Hz, 1H), 6.77 (d, *J* = 0.5 Hz, 1H), 2.51 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 140.7 (s), 138.7 (s), 137.5 (s), 132.7 (s), 130.0 (s), 129.2 (s), 128.8 (s), 128.5 (s), 128.1 (s), 128.0 (s), 127.2 (s), 123.9 (s), 120.2 (s), 110.3 (s), 103.3 (s), 21.4 (s).

5-Fluoro-1,2-diphenyl-1H-indole (**3***j*). Yield 46 mg (81%) of a white solid; 103–105 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.49–7.43 (m, 2H), 7.42–7.35 (m, 2H), 7.31–7.23 (m, 8H), 6.96 (td, *J* = 9.1, 2.5 Hz, 1H), 6.80 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 158.4 (d, *J*_{C-F} = 235.4 Hz), 142.2 (s), 138.3 (s), 135.6 (s), 132.2 (s), 129.3 (s), 128.9 (s), 128.5 (d, *J*_{C-F} = 10.3 Hz), 128.2 (s), 127.9 (s), 127.5 (s), 127.4 (s), 111.3 (d, *J*_{C-F} = 9.6 Hz), 110.5 (d, *J*_{C-F} = 26.1 Hz), 105.2 (d, *J*_{C-F} = 23.6 Hz), 103.5 (d, *J*_{C-F} = 4.5 Hz); HRMS (ESI) *m*/*z* calcd for C₂₀H₁₄FN [M + H]⁺ 288.1189, found 288.1185.

1-Phenyl-2-(thiophen-3-yl)-1H-indole (**3**).^{16b} Yield 44 mg (81%) of a white solid; mp 131–133 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.73–7.67 (m, 1H), 7.55–7.44 (m, 3H), 7.39–7.34 (m, 2H), 7.26–7.17 (m, 4H), 7.07 (dd, J = 5.0, 1.2 Hz, 1H), 6.90–6.85 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 139.0 (s), 138.5 (s), 136.0 (s), 133.0 (s), 129.4 (s), 128.4 (s), 128.0 (s), 127.8 (s), 127.8 (s), 125.0 (s), 122.3 (s), 122.2 (s), 120.6 (s), 120.4 (s), 110.4 (s), 102.6 (s).

1,3-Bis(1-phenyl-1H-indol-2-yl)benzene (**3m**). Yield 74 mg (81%) of a yellow solid; mp 141–143 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.73–7.67 (m, 2H), 7.48 (dd, J = 10.0, 4.7 Hz, 4H), 7.44–7.40 (m, 2H), 7.34–7.31 (m, 3H), 7.29 (s, 1H), 7.27–7.20 (m, 8H), 7.11 (t, J = 1.8 Hz, 2H), 6.60 (d, J = 0.4 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 140.2 (s), 138.9 (s), 138.4 (s), 132.4 (s), 129.5 (s), 129.3 (s), 127.9 (s), 127.7 (s), 127.3 (s), 122.4 (s), 120.7 (s), 120.5 (s), 110.6 (s), 103.7 (s); HRMS (ESI) m/z calcd for C₃₄H₂₄N₂ [M + H]⁺: 461.2018, found 461.2014.

2-Phenyl-1-(p-tolyl)-1H-indole (**3ba**).^{16c} Yield 45 mg (80%) of a yellow solid; mp 78–80 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.73 (dd, J = 6.0, 2.9 Hz, 1H), 7.34–7.16 (m, 12H), 6.84 (s, 1H), 2.44 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 140.7 (s), 139.1 (s), 137.0 (s), 135.8 (s), 132.6 (s), 129.9 (s), 128.9 (s), 128.2 (s), 128.1 (s), 127.8 (s), 127.2 (s), 122.2 (s), 120.5 (s), 120.4 (s), 110.7 (s), 103.4 (s), 21.1 (s).

1-(4-(tert-Butyl)phenyl)-2-phenyl-1H-indole (**3ca**).^{17a} Yield 54 mg (84%) of a white solid; mp 132–134 °C; ¹H NMR(400 MHz, CDCl₃) δ 7.73 (dd, *J* = 6.2, 2.9 Hz, 1H), 7.45 (d, *J* = 8.5 Hz, 2H), 7.37–7.31 (m, 3H), 7.30–7.26 (m, 3H), 7.23–7.19 (m, 4H), 6.84 (s, 1H), 1.40 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 150.2 (s), 140.7 (s), 139.1 (s), 135.7 (s), 132.6 (s), 128.9 (s), 128.1 (s), 128.1 (s), 127.4 (s), 127.2 (s), 126.1 (s), 122.1 (s), 120.5 (s), 120.4 (s), 110.8 (s), 103.4 (s), 34.6 (s), 31.4 (s).

1-(4-Chlorophenyl)-2-phenyl-1H-indole (**3da**).^{17b} Yield 45 mg (75%) of a yellow solid; mp 105–107 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.73–7.69 (m, 1H), 7.43–7.39 (m, 2H), 7.29 (d, *J* = 3.0 Hz, 6H), 7.24–7.20 (m, 4H), 6.83 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 140.6 (s), 138.8 (s), 137.1 (s), 132.9 (s), 132.2 (s), 130.9 (s), 129.5 (s), 129.2 (s), 128.9 (s), 128.3 (s), 128.3 (s), 127.5 (s), 122.6 (s), 122.2 (s), 121.5 (s), 120.9 (s), 120.7 (s), 110.4 (s), 104.1 (s).

1-(4-Bromophenyl)-2-phenyl-1H-indole (**3ea**). Yield 48 mg (69%) of a yellow solid; mp 111–114 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.74–7.70 (m, 1H), 7.59–7.54 (m, 2H), 7.30 (t, J = 4.8 Hz, 6H), 7.24–7.21 (m, 2H), 7.18–7.14 (m, 2H), 6.83 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 140.6 (s), 138.8 (s), 137.6 (s), 132.5 (s), 132.2 (s), 129.5 (s), 128.9 (s), 128.4 (s), 128.3 (s), 127.5 (s), 122.6 (s), 121.0 (s), 120.8 (s), 120.7 (s), 110.4 (s), 104.2 (s); HRMS (ESI) *m/z* calcd for C₂₀H₁₄BrN [M + H]⁺ 348.0388, found 348.0382.

1-(4-Fluorophenyl)-2-phenyl-1H-indole (**3fa**).^{14a} Yield 44 mg (79%) of a white solid; mp 99–101 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.85–7.76 (m, 1H), 7.41–7.30 (m, 10H), 7.20 (t, *J* = 8.5 Hz, 2H), 6.93 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 161.4 (d, *J*_{C-F} = 247.2 Hz), 140.7 (s), 139.1 (s), 134.5 (d, *J*_{C-F} = 3.1 Hz), 132.3 (s), 129.6 (d, *J*_{C-F} = 8.5 Hz), 129.1 (s), 128.9 (s), 128.2 (s), 127.4 (s), 122.4 (s), 121.5 (s), 120.6 (s), 116.2 (d, *J*_{C-F} = 22.7 Hz), 110.4 (s), 103.7 (s).

1-(4-lodophenyl)-2-phenyl-1H-indole (**3ga**). Yield 51 mg (65%) of a yellow solid; mp 134–136 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.75–7.69 (m, 1H), 7.48–7.41 (m, 2H), 7.38 (dt, *J* = 9.6, 4.3 Hz, 1H), 7.31–7.26 (m, 8H), 7.21 (dd, *J* = 6.1, 3.1 Hz, 2H), 6.83 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 140.7 (s), 139.0 (s), 138.5 (s), 132.5 (s), 129.2 (s), 128.9 (s), 128.2 (s), 128.1 (s), 128.0 (s), 127.3 (s), 127.2 (s), 122.3 (s), 120.7 (s), 120.5 (s), 110.6 (s), 103.7 (s); HRMS (ESI) *m/z* calcd for C₂₀H₁₄IN [M + H]⁺ 396.0249, found 396.0238.

1-(*p*-Ethoxycarbonylphenyl)-2-phenyl-4-1H-indole (**3ha**).^{18a} Yield 52 mg (77%) of a white solid; mp 136–138 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.12 (d, *J* = 8.5 Hz, 2H), 7.74–7.69 (m, 1H), 7.34 (d, *J* = 8.5 Hz, 2H), 7.28(d, *J* = 3.5 Hz, 7H), 7.23 (dd, *J* = 4.4, 2.2 Hz, 1H), 6.85 (s, 1H), 4.42 (q, *J* = 7.1 Hz, 2H), 1.43 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 166.0 (s), 142.5 (s), 140.5 (s), 138.5 (s), 132.2 (s), 130.6 (s), 128.9 (s), 128.9 (s), 128.5 (s), 128.3 (s), 127.6 (s), 127.5 (s), 122.7 (s), 121.1 (s), 120.7 (s), 110.4 (s), 104.8 (s), 61.2 (s), 14.3 (s).

2-Phenyl-1-(o-tolyl)-1H-indole (**3ia**).^{14b} Yield 40 mg (71%) of a yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 7.75–7.71(m, 1H), 7.39–7.34 (m, 1H), 7.34–7.27 (m, 6H), 7.25 (qd, *J* = 4.6, 1.7 Hz, 3H), 7.19 (dd, *J* = 6.4, 2.7 Hz, 1H), 6.98 (dd, *J* = 5.9, 2.5 Hz, 1H), 6.89–6.84 (m, 1H), 1.90 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 141.1 (s), 139.0 (s), 137.6 (s), 136.9 (s), 132.7 (s), 131.2 (s), 129.5 (s), 129.2 (s), 128.4 (s), 128.2 (s), 128.2 (s), 127.5 (s), 127.3 (s), 126.8 (s), 122.2 (s), 120.4 (s), 110.8 (s), 102.6 (s), 17.6 (s).

1-(3-Bromophenyl)-2-phenyl-1H-indole (**3***ja*).^{14b} Yield 53 mg (76%) of a yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 7.73–7.68 (m, 1H), 7.54–7.48 (m, 2H), 7.29 (d, *J* = 3.5 Hz, 8H), 7.23 (dd, *J* = 6.4, 2.7 Hz, 2H), 7.14 (d, *J* = 7.7 Hz, 1H), 6.83 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 140.6 (s), 139.9 (s), 138.8 (s), 132.1 (s), 130.9 (s), 130.4 (s), 130.3 (s), 128.9 (s), 128.4 (s), 128.3 (s), 127.6 (s), 126.8 (s), 122.6 (s), 122.5 (s), 121.1 (s), 120.7 (s), 110.4 (s), 104.3 (s).

Preparation of Bis(1,2-diphenyl-1*H*-indol-3-yl)methanes 4. 1,2-Diphenyl-1*H*-indoles 3 (0.05 mmol), $Pd(OAc)_2$ (0.1 equiv), and AgOAc (1 equiv) were added in DMSO (1 mL) in a round-bottomed flask under N₂ and stirred at 130 °C for 20 h until TLC indicated complete consumption. The reaction mixture was diluted with water and extracted with ethyl acetate. Then the combined organic layer was evaporated under reduced pressure and the residue was purified by chromatography (silica gel, eluent EtOAc/PE = 1/100) to give 4.

Bis(1,2-diphenyl-1H-indol-3-yl)methane (4a). Yield 12 mg (89%) of a white solid; mp 149–151 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.34 (dd, J = 14.7, 7.1 Hz, 6H), 7.29–7.24 (m, 9H), 7.22–7.16 (m, 7H), 7.14–7.10 (m, 2H), 6.98 (dd, J = 11.1, 3.9 Hz, 2H), 4.52 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 138.5 (s), 137.7 (s), 137.2 (s), 132.2 (s), 130.9 (s), 128.9 (s), 128.5 (s), 128.0 (s), 127.8 (s), 127.2 (s), 126.5 (s), 122.1 (s), 119.9 (s), 119.8 (s), 113.9 (s), 110.1 (s), 21.2 (s); HRMS (ESI) m/z calcd for C₄₁H₃₀N₂ [M]⁺ 550.2409, found 550.2404.

Bis(2-*phenyl*-1-(*p*-tolyl)-1*H*-*indol*-3-yl)*methane* (**4b**). Yield 13 mg (90%) of a yellow solid; mp 153–156 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.29 (d, *J* = 6.1 Hz, 4H), 7.22 (dt, *J* = 13.3, 2.9 Hz, 10H), 7.14–7.02 (m, 10H), 6.95 (t, *J* = 7.5 Hz, 2H), 4.48 (s, 2H), 2.36 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 137.8 (s), 137.2 (s), 136.3 (s), 135.9 (s), 132.3 (s), 130.9 (s), 129.5 (s), 128.4 (s), 127.8 (s), 127.7 (s), 127.1 (s), 121.9 (s), 119.8 (s), 119.6 (s), 113.7 (s), 110.2 (s), 21.2 (s), 21.1 (s); HRMS (ESI) *m*/*z* calcd for C₄₃H₃₄N₂ [M]⁺ 578.2722, found 578.2719.

Bis(1-(4-(tert-butyl)phenyl)-2-phenyl-1H-indol-3-yl)methane (4c). Yield 15 mg (88%) of a yellow solid; mp 155–157 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.33 (d, *J* = 8.5 Hz, 3H), 7.31–7.27 (m, 5H), 7.25– 7.20 (m, 10H), 7.09 (dd, J = 13.0, 4.7 Hz, 6H), 6.95 (t, J = 7.2 Hz, 2H), 4.49 (s, 2H), 1.33 (s, 18H); ¹³C NMR (100 MHz, CDCl₃) δ 149.4 (s), 137.8 (s), 137.2 (s), 135.8 (s), 132.3 (s), 130.9 (s), 128.4 (s), 127.7 (s), 127.4 (s), 127.1 (s), 125.7 (s), 121.9 (s), 119.8 (s), 119.6 (s), 113.7 (s), 110.3 (s), 34.5 (s), 31.3 (s), 21.2 (s); HRMS (ESI) m/z calcd for C₄₉H₄₆N₂ [M]⁺ 662.3661, found 662.3657.

Bis(5-methyl-1,2-diphenyl-1H-indol-3-yl)methane (4d). Yield 12 mg (80%) of a yellow solid; mp 150–152 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.29 (dd, *J* = 15.3, 9.6 Hz, 16H), 7.16–7.06 (m, 6H), 6.93–6.82 (m, 4H), 4.50 (s, 2H), 2.24 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 138.8 (s), 137.0 (s), 136.1 (s), 132.6 (s), 131.0 (s), 128.8 (s), 128.6 (s), 128.0 (s), 127.9 (s), 127.2 (s), 126.3 (s), 123.6 (s), 119.7 (s), 113.9 (s), 109.7 (s), 21.2 (s), 20.6 (s); HRMS (ESI) *m*/*z* calcd for C₄₃H₃₄N₂ [M]⁺ 578.2722, found 578.2718.

Bis(1-(4-fluorophenyl)-2-phenyl-1H-indol-3-yl)methane (4e). Yield 9 mg (64%) of a yellow solid; mp 161–163 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.30–7.24 (m, 9H), 7.19–7.09 (m, 11H), 7.05–6.95 (m, 6H), 4.47 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 161.0 (d, $J_{C-F} = 246.5$ Hz), 137.8 (s), 137.2 (s), 134.5 (d, $J_{C-F} = 3.0$ Hz), 131.9 (s), 130.9 (s), 129.6 (d, $J_{C-F} = 8.5$ Hz), 128.4 (s), 127.9 (s), 127.3 (s), 122.2 (s), 119.9 (d, $J_{C-F} = 8.8$ Hz), 115.8 (d, $J_{C-F} = 22.7$ Hz), 113.8 (s), 109.9 (s), 21.1 (s); HRMS (ESI) *m*/*z* calcd for C₄₁H₂₈N₂F₂ [M]⁺ 586.2221, found 586.2216.

Bis(1-(4-chlorophenyl)-2-phenyl-1H-indol-3-yl)methane (4f). Yield 10 mg (67%) of a yellow solid; mp 164–167 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.30 (d, J = 10.1 Hz, 9H), 7.26 (d, J = 2.5 Hz, 4H), 7.21 (d, J = 8.2 Hz, 2H), 7.16–7.11 (m, 5H), 7.10–7.05 (m, 4H), 6.98 (dd, J = 11.0, 3.9 Hz, 2H), 4.46 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 137.5 (s), 137.1 (s), 137.0 (s), 132.2 (s), 131.8 (s), 130.8 (s), 129.1 (s), 129.1 (s), 128.6 (s), 128.0 (s), 127.4 (s), 122.3 (s), 120.1 (s), 119.9 (s), 114.2 (s), 109.9 (s), 21.0 (s); HRMS (ESI) m/z calcd for C₄₁H₂₈N₂Cl₂ [M]⁺ 618.1630, found 618.1624.

Bis(1,2-*diphenyl*-1*H*-*indol*-3-*yl*)*methane* (**4aa**). Yield 12 mg (86%) of a white solid; mp 151–153 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.37–7.31 (m, 5H), 7.30–7.22 (m, 12H), 7.21–7.15 (m, 7H), 7.13–7.08 (m, 2H), 6.98 (dd, *J* = 11.0, 3.9 Hz, 2H); HRMS (ESI) *m/z* calcd for C₄₁H₂₈D₂N₂ [M]⁺ 552.2535, found 552.2528.

Bis(1,2-diphenyl-1H-indol-3-yl)-l2-methane-¹³C (4ab). Yield 11 mg (83%) of a white solid; mp 150–153 °C; ¹³C NMR (100 MHz, CDCl₃) δ 138.5 (s),137.7 (s), 137.7 (s), 132.2 (s), 130.9 (s), 128.9 (s), 128.5 (s), 128.0 (s), 127.8 (s), 127.2 (s), 126.5 (s), 122.1 (s), 119.8 (s), 119.8 (s), 110.1 (s), 21.2 (s); HRMS (ESI) *m*/*z* calcd for C_{40}^{13} CH₃₀N₂ [M]⁺ 551.2443, found 551.2437.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b00106.

¹H and ¹³C NMR spectra for all compounds, ¹H NMR spectrum of reaction of **1b** with **2a**, and of ESI-(+)-MS data (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: cuixl@hqu.edu.cn.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This work was supported by NSF of China (21572072), Xiamen Southern Oceanographic Center (15PYY052SF01), Science and Technology Bureau of Xiamen City (3502Z20150054), and Huaqiao University.

The Journal of Organic Chemistry

REFERENCES

(1) (a) Sundberg, R. J. *Indoles*; Academic Press: San Deigo, CA, 1996.
(b) Horton, D. A.; Bourne, G. T.; Smythe, M. L. *Chem. Rev.* 2003, 103, 893.
(c) Somei, M.; Yamada, F. *Nat. Prod. Rep.* 2005, 22, 73.
(d) Higuchi, K.; Kawasaki, T. *Nat. Prod. Rep.* 2007, 24, 843.
(e) Kochanowska-Karamyan, A. J.; Hamann, M. T. *Chem. Rev.* 2010, 110, 4489.
(f) Inman, M.; Moody, C. J. *Chem. Sci.* 2013, 4, 29.

(2) (a) Zeni, G.; Larock, R. C. Chem. Rev. 2004, 104, 2285.
(b) Cacchi, S.; Fabrizi, G. Chem. Rev. 2005, 105, 2873. (c) Humphrey, G. R.; Kuethe, J. T. Chem. Rev. 2006, 106, 2875. (d) Ackermann, L. Synlett 2007, 2007, 0507. (e) Patil, N. T.; Yamamoto, Y. Chem. Rev. 2008, 108, 3395. (f) Bandini, M.; Eichholzer, A. Angew. Chem., Int. Ed. 2009, 48, 9608. (g) Cacchi, S.; Fabrizi, G. Chem. Rev. 2011, 111, PR215. (h) Patureau, F. W.; Glorius, F. Angew. Chem., Int. Ed. 2011, 50, 1977. (i) Shiri, M. Chem. Rev. 2012, 112, 3508.

(3) (a) Fischer, E.; Jourdan, F. Ber. Dtsch. Chem. Ges. 1883, 16, 2241.
(b) Robinson, B. The Fischer Indole Synthesis; John Wiley & Sons Inc.: New York, 1982. (c) El Kaim, L.; Grimaud, L.; Ronsseray, C. Synlett 2010, 2010, 2296.

(4) (a) Stevens, C. V.; Van Meenen, E.; Eeckhout, Y.; Vanderhoydonck, B.; Hooghe, W. Chem. Commun. 2005, 38, 4827.
(b) Batail, N. L.; Bendjeriou, A.; Djakovitch, L.; Dufaud, V. Appl. Catal, A 2010, 388, 179. (c) Van derJeught, S.; De Vos, N.; Masschelein, K.; Ghiviriga, I.; Stevens, C. V. Eur. J. Org. Chem. 2010, 2010, 5444.

(5) (a) Stuart, D. R.; Bertrand-Laperle, M.; Burgess, K. M. N.; Fagnou, K. J. Am. Chem. Soc. **2008**, 130, 16474. (b) Houlden, C. E.; Bailey, C. D.; Ford, J. G.; Gagné, M. R.; Lloyd-Jones, G. C.; Booker-Milburn, K. I. J. Am. Chem. Soc. **2008**, 130, 10066. (c) Terao, Y.; Wakui, H.; Satoh, T.; Miura, M.; Nomura, M. J. Am. Chem. Soc. **2001**, 123, 10407.

(6) (a) Bedford, R. B.; Coles, S. J.; Hursthouse, M. B.; Limmert, M. E. Angew. Chem., Int. Ed. 2003, 42, 112. (b) Kawatsura, M.; Hartwig, J. F. J. Am. Chem. Soc. 1999, 121, 1473. (c) Fox, J. M.; Huang, X.; Chieffi, A.; Buchwald, S. L. J. Am. Chem. Soc. 2000, 122, 1360.

(7) (a) Kakiuchi, F.; Matsuura, Y.; Kan, S.; Chatani, N. J. Am. Chem. Soc. 2005, 127, 5936. (b) Kalyani, D.; Deprez, N. R.; Desai, L. V.; Sanford, M. S. J. Am. Chem. Soc. 2005, 127, 7330. (c) Gao, J. L.; Shao, Y. Y.; Zhu, J. Y.; Zhu, J. Q.; Mao, H.; Wang, X. X.; Lv, X. J. Org. Chem. 2014, 79, 9000.

(8) (a) Yusubov, M. S.; Maskaev, A. V.; Zhdankin, V. V. ARKIVOC
2011, 370. (b) Merritt, E. A.; Olofsson, B. Angew. Chem., Int. Ed. 2009, 48, 9052. (c) Zhdankin, V. V.; Stang, P. J. Chem. Rev. 2008, 108, 5299. (d) Grushin, V. V. Chem. Soc. Rev. 2000, 29, 315. (e) Stang, P. J.; Zhdankin, V. V. Chem. Rev. 1996, 96, 1123. (f) Chen, B.; Hou, X. L.; Li, Y. X.; Wu, Y. D. J. Am. Chem. Soc. 2011, 133, 7668. (g) Deprez, N. R.; Sanford, M. S. J. Am. Chem. Soc. 2009, 131, 11234. (h) Phipps, R. J.; Grimster, N. P.; Gaunt, M. J. J. Am. Chem. Soc. 2008, 130, 8172. (i) Peng, J.; Chen, C.; Wang, Y.; Lou, Z.; Li, M.; Xi, C.; Chen, H. Angew. Chem., Int. Ed. 2013, 52, 7574.

(9) Li, P. F.; Cheng, G. L.; Zhang, H.; Xu, X. X.; Gao, J. Y.; Cui, X. L. J. Org. Chem. **2014**, 79, 8156.

(10) (a) Yan, L. P.; Zhao, D. B.; Lan, J. B.; Cheng, Y. Y.; Guo, Q.; Li,

X. Y.; Wu, N. J.; You, J. S. Org. Biomol. Chem. 2013, 11, 7966.
(b) Takeda, D.; Hirano, K.; Satoh, T.; Miura, M. Heterocycles 2012, 86, 487.

(11) Baik, C.; Kim, D.; Kang, M.-S.; Song, K.; Kang, S. O.; Ko, J. *Tetrahedron* **2009**, *65*, 5302.

(12) (a) Brand, J. P.; Chevalley, C.; Waser, J. Beilstein J. Org. Chem. 2011, 7, 565. (b) Swamy, N. K.; Yazici, A.; Pyne, S. G. J. Org. Chem. 2010, 75, 3412.

(13) Bielawski, M.; Zhu, M. Z.; Olofsson, B. Adv. Synth. Catal. 2007, 349, 2610.

(14) (a) Fang, Y. – Q.; Lautens, M. J. Org. Chem. 2008, 73, 538.
(b) Gao, J. L.; Shao, Y. Y.; Zhu, J. Y.; Zhu, J. Q.; Mao, H.; Wang, X. X.; Lv, X. J. Org. Chem. 2014, 79, 9000.

(15) (a) Willis, M. C.; Brace, G. N.; Findlay, T. J. K.; Holmes, I. P. Adv. Synth. Catal. 2006, 348, 851. (b) Fang, Y. - Q.; Lautens, M. Org.

Lett. 2005, 7, 3549. (c) Sharma, U.; Kancherla, R.; Naveen, T.; Agasti, S.; Maiti, D. Angew. Chem. 2014, 126, 12089.

(16) (a) Barluenga, J.; Jimenez-Aquino, A.; Aznar, F.; Valdes, C. J. Am. Chem. Soc. 2009, 131, 4031. (b) Crawford, S. M.; Lavery, C. B.; Stradiotto, M. Chem. - Eur. J. 2013, 19, 16760. (c) Kaspar, L. T.; Ackermann, L. Tetrahedron 2005, 61, 11311.

(17) (a) Old, D. W.; Harris, M. C.; Buchwald, S. L. Org. Lett. 2000, 2, 1403. (b) Barluenga, J.; Jimenez-Aquino, A.; Aznar, F.; Valdes, C. Chem. - Eur. J. 2010, 16, 11707.

(18) (a) Ackermann, L. Org. Lett. 2005, 7, 439. (b) Sharma, U.; Kancherla, R.; Naveen, T.; Agasti, S.; Maiti, D. Angew. Chem., Int. Ed. 2014, 53, 11895.